

# Damage Free and Energy Efficient Megasonic Cleaning

## PIs:

- **Srini Raghavan, Materials Science and Engineering, UA**
- **Manish Keswani, Materials Science and Engineering, UA**

## Graduate Students:

- **Zhenxing Han: PhD candidate, Chemical and Environmental Engineering, UA**

## Core Funding Source:

- **Intel (mentors: Gopal Rao and Avi Fuerst)**

## Cost Share:

- **ERC (~ \$25k)**
- **IMEC: In kind donation of Patterned Wafers (~\$5k)**

# Objectives

## **Overall Objective:**

- **Energy reduction in Megasonic Cleaning Processes**

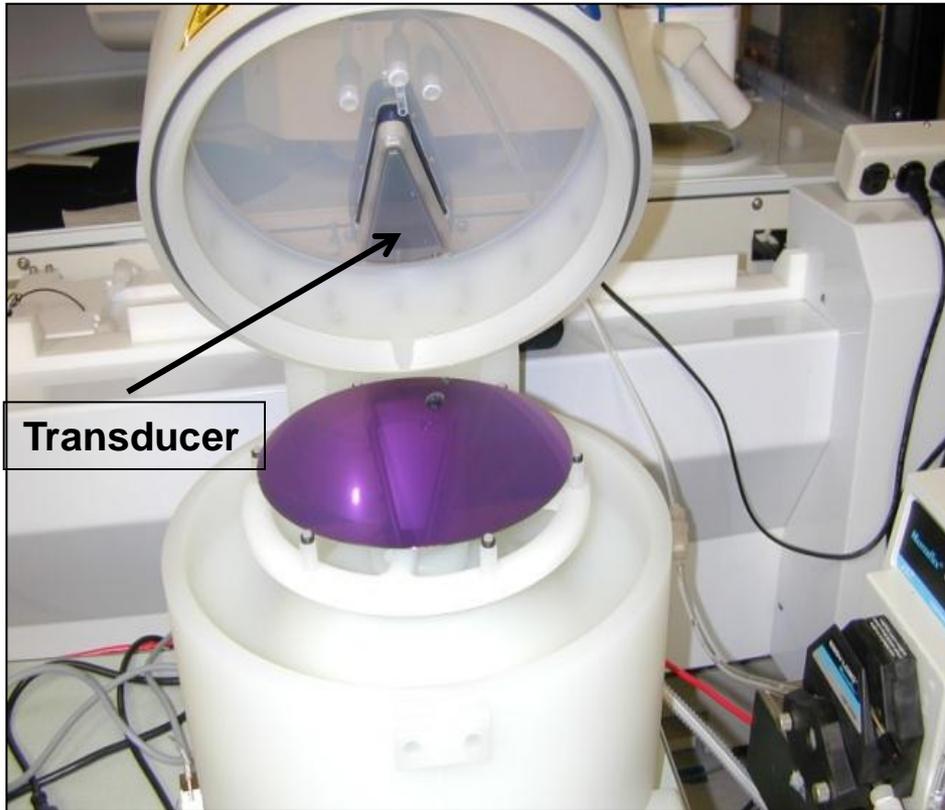
## **Current Year Goal:**

- **Development of a Chemical System for a Megasonic Cleaning Process that is able to achieve high particle removal efficiency (PRE) at low power density with significantly reduced feature damage**

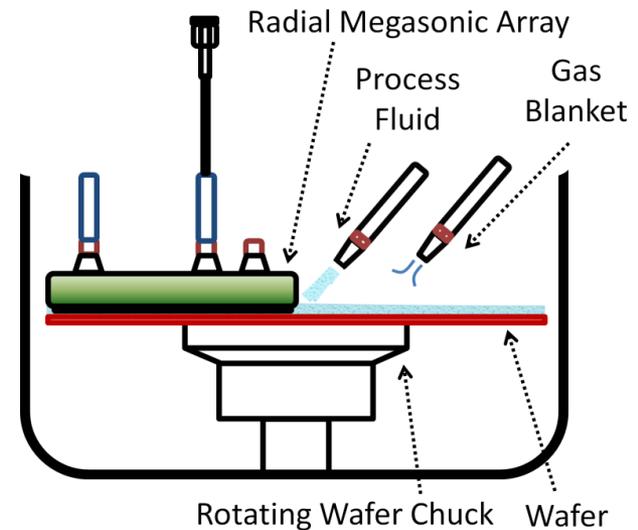
# ESH Metrics and Impact

- 1. Cleaning formulations developed in this work use weakly alkaline pH (~8.2) to achieve effective particle removal as opposed to traditional cleaning formulations based on ammonium-peroxide mixtures (APM) that operate under more alkaline (pH > 10) conditions**
- 2. Overall die/wafer yield improvement through reduction in feature damage by proper choice of cleaning chemicals and low transducer power density (energy)**

# Single Wafer MegPie<sup>®</sup> Set up for Cleaning and Damage Studies

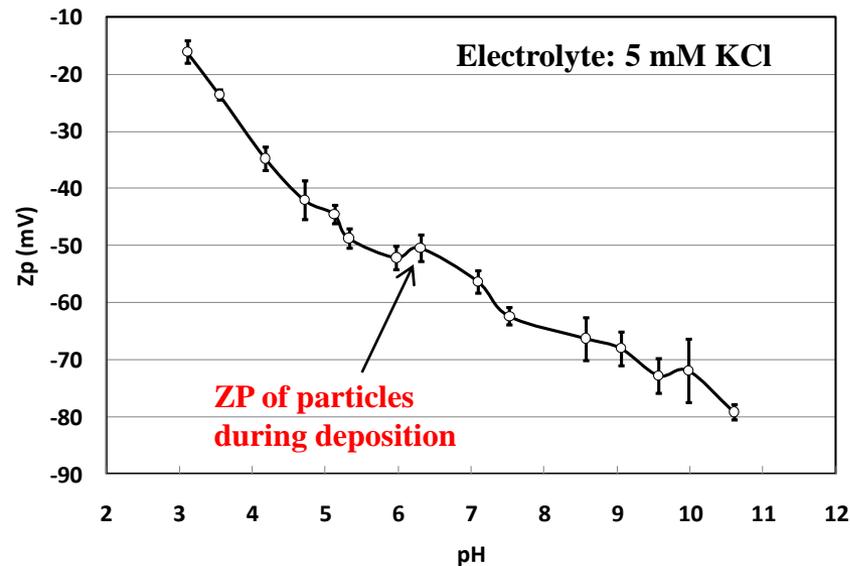


- Frequency ~ **925 kHz**
- Power density Range = **0.15 - 2.9 W/cm<sup>2</sup>**
- Active megasonic surface area = **32.3 cm<sup>2</sup>**
- Substrate to piezoelectric transducer distance ~ **1 mm**



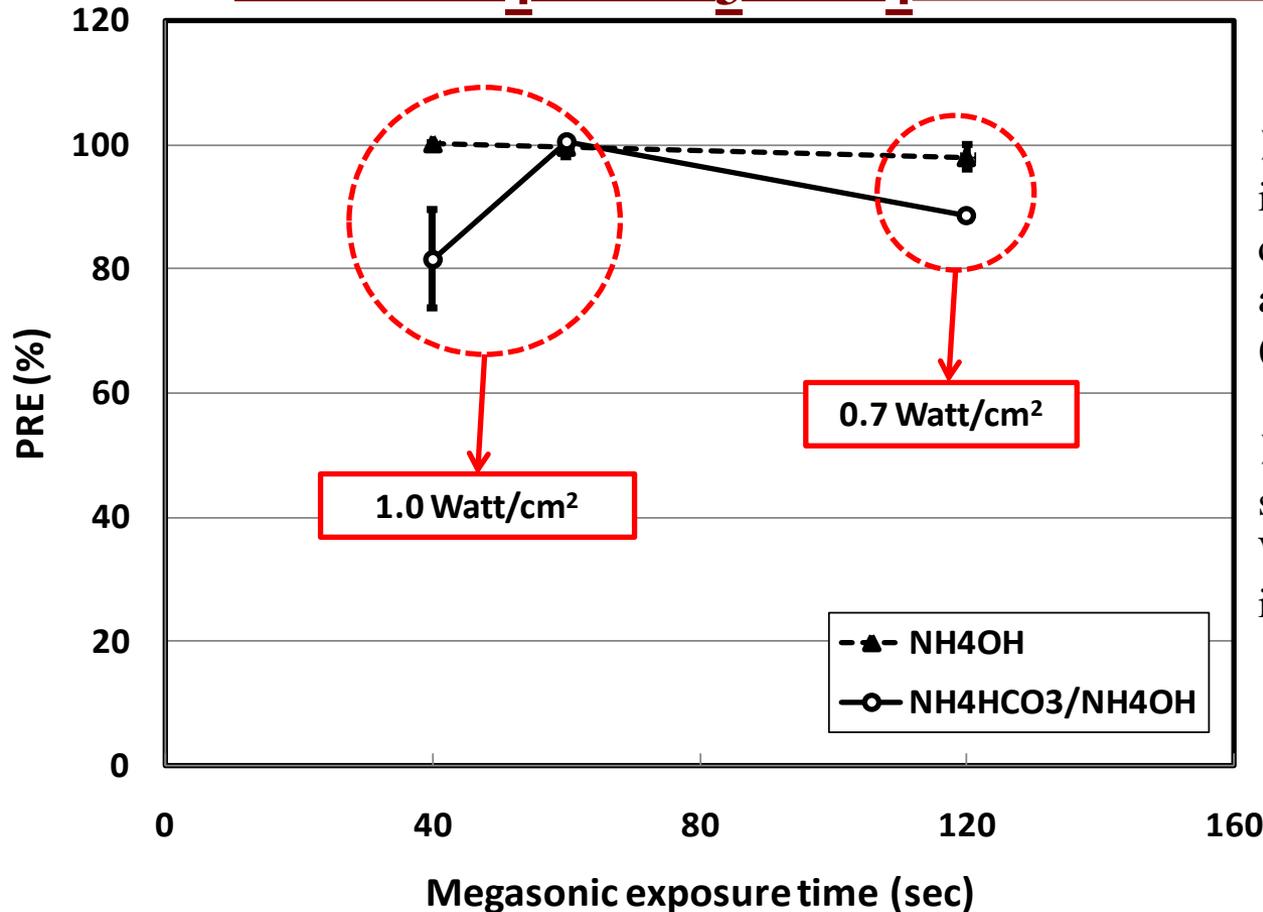
# Procedure for Cleaning Study on Blanket Wafers

1. 8" oxide wafers pre-cleaned using dilute HF (1:100)
2. Silica particles (mean diameter ~200 nm) deposited on the wafer using spin deposition technique
3. ~4000 particles deposited per wafer and aged for 24 hr before cleaning
4. The contaminated oxide wafers cleaned in MegPie<sup>®</sup> under the following conditions:
  - Cleaning solution flow rate: 300 ml/min, wafer spin speed: 30 rpm.
  - Megasonic conditions: Power density: 0.7-1.0 W/cm<sup>2</sup>, Exposure time: 40, 60 and 120 sec
5. Particle count measured using SP1 after deposition and cleaning



**Zeta potential measurement of particles  
in the silica slurry**

# Cleaning Results on Blanket Wafers Using $\text{NH}_4\text{OH}$ and $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$ solutions at pH 8.2



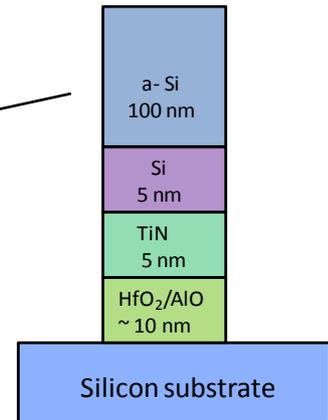
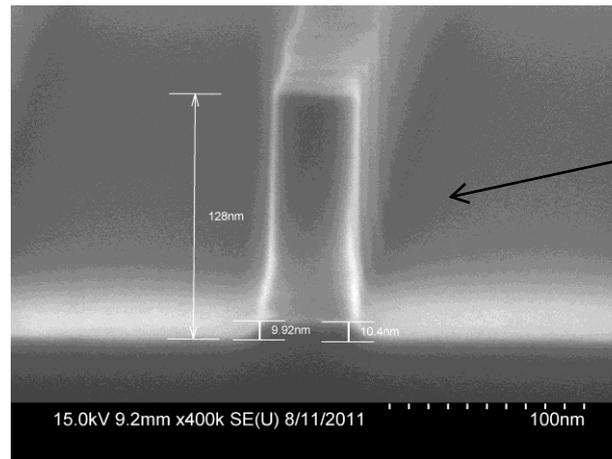
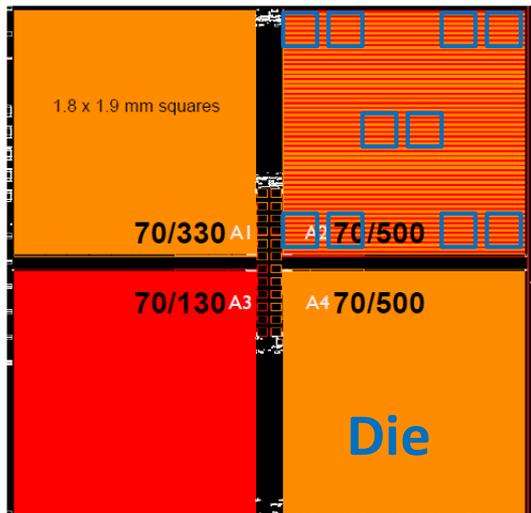
➤ In  $\text{NH}_4\text{OH}$  solution, 100 % PRE is achieved for investigated power densities (0.7 and 1.0  $\text{W/cm}^2$ ) and megasonic exposure times (40, 60 and 120 sec)

➤ In  $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$  solutions, slightly higher power density of 1.0  $\text{W/cm}^2$  and cleaning time of 60 sec is required to achieve 100 % PRE

**Proper choice of power density and cleaning time yields a 100 % PRE in  $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$  solution (at pH of ~ 8.2)**

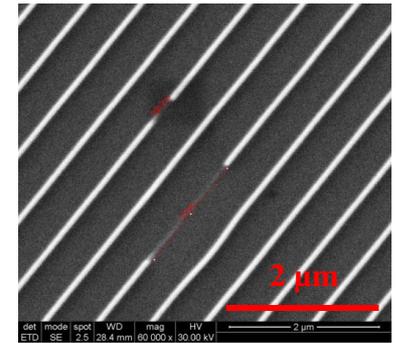
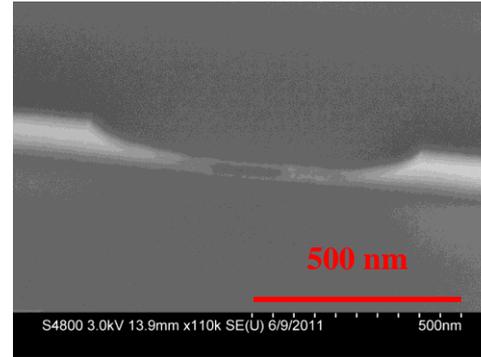
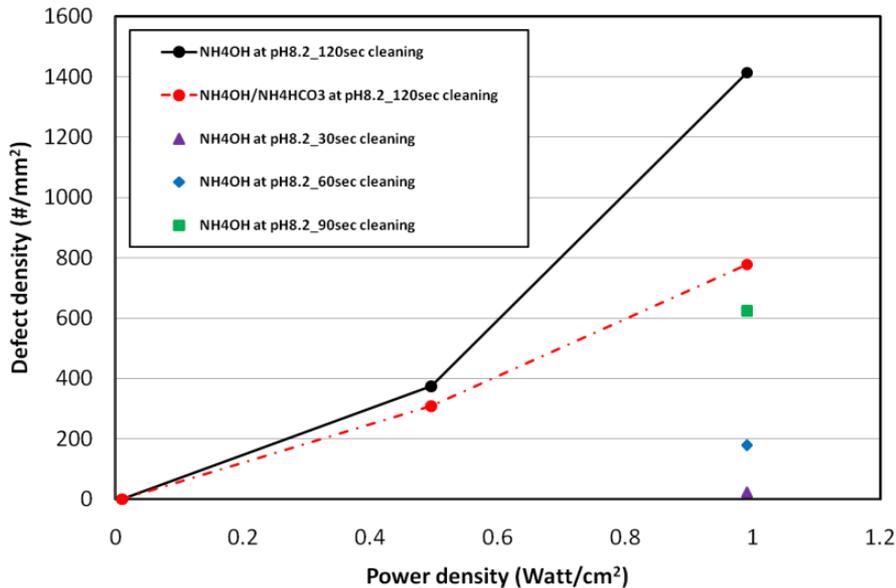
# Procedure for Cleaning and Damage Study on Patterned Samples

- Patterned wafers (high-k-metal gate) obtained from IMEC. The A2 section (1.8 mm x 1.9 mm) in the structure, which consisted of lines with 40 nm width (measured by SEM) separated by 500 nm, used for characterization.
- Leica DM4000 M microscope used for imaging; 10 regions each of 0.3 mm x 0.4 mm area imaged at a magnification of 200X.
- The images were then processed using the ImageJ software. It was established earlier that by adjusting suitable threshold value to the image, it is capable to distinguish between particles and defects.



SEM showing multi-stack line on Si substrate

# Effect of Power Density and Megasonic Exposure Time on Particle Removal and Feature Damage in $\text{NH}_4\text{OH}$ and $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$ Solutions



**SEM image of typical defects on lines during megasonic exposure**

Cleaning Solution	PRE (%) Power density of 1 W/cm <sup>2</sup> and meg exposure for 60 sec
$\text{NH}_4\text{OH}$	~ 95
$\text{NH}_4\text{OH}/\text{NH}_4\text{HCO}_3$	~ 94

➤ **High PRE achieved in both  $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$  &  $\text{NH}_4\text{OH}$  systems at 1 W/cm<sup>2</sup> and 60 sec (meg exposure)**

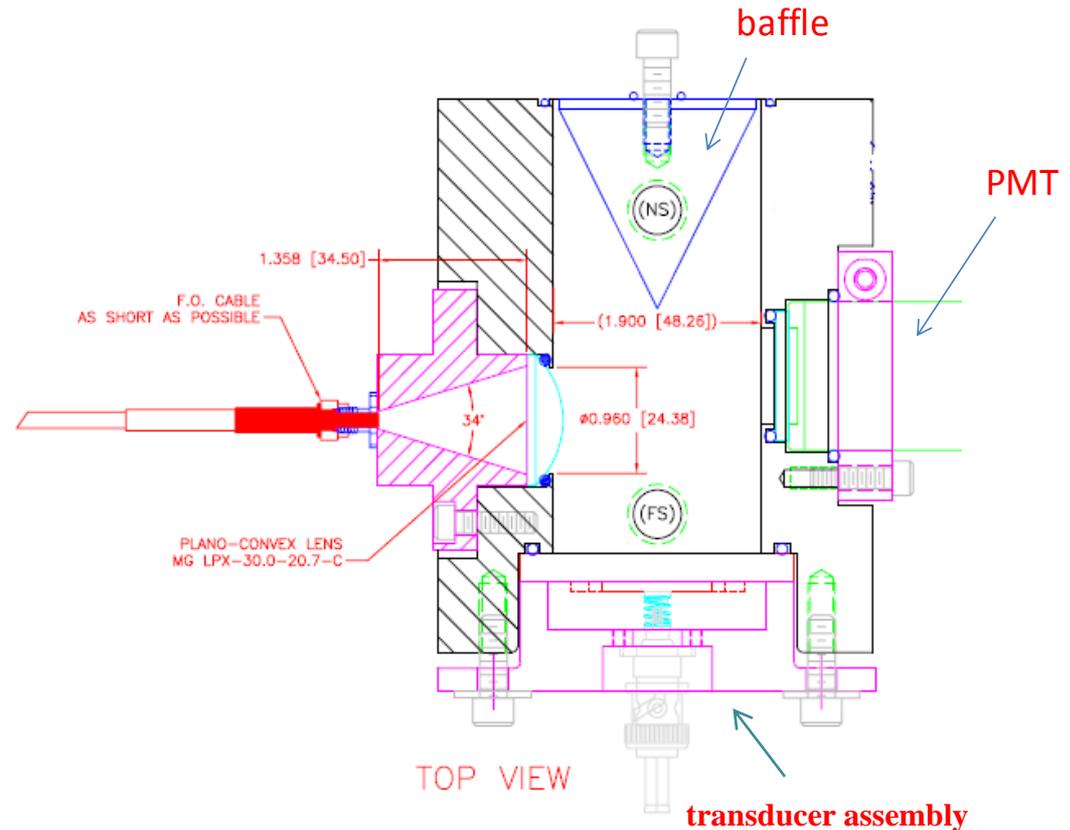
➤ **Damage to thin lines much higher in  $\text{NH}_4\text{OH}$  solution (pH ~ 8.2) compared to that in  $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$  solutions at pH 8.2 for 120 sec of meg exposure; Defect study in  $\text{NH}_4\text{HCO}_3/\text{NH}_4\text{OH}$  system at lower meg exposure time in progress**

# Current Status on the CT Cell-Spectrometer Assembly

## QE65000/FO cable Specifications

Wavelength range	200-950 nm
Spectral Resolution	~ 5 nm
Integration time	8 ms – 15 min
FO cable	1000 $\mu\text{m}$ UV fiber

To spectrometer  
QE65000 (Ocean  
Optics)



A plano-convex lens (of diameter 3 cm) has been installed in the CT cell (opposite to the photomultiplier tube) to have better light collection efficiency

Tests are being conducted to determine light focusing and transmission capability of the optical set-up

# **Industrial Interactions and Technology Transfer**

- **Collaboration with ProSys, CA and technical discussions with Ocean Optics for design and construction of the optical set-up for integration of the spectrometer with the CT cell**
- **Filed an Invention Disclosure on 10/07/11 with University of Arizona on a novel chemical formulation developed for low damage and high PRE megasonic cleaning process. Ongoing discussions with Intel on filing a patent**

## **Acknowledgements**

- **Dr. Sangita Kumari, Global Foundries, who worked during the first phase of the project**
- **Mark Beck and Eric Liebscher, ProSys, for help with modification of the CT cell and optical assembly**

# Future Plans

## Next Year Plans

- **Collect sonoluminescence spectra in ammonium bicarbonate based chemical system that has shown good cleaning with low feature damage**
- **Relate damage to high energy cavitation events during megasonic cleaning through spectral information**

# Publications, Presentations, and Recognitions/Awards

- **S. Kumari, M. Keswani, S. Singh, M. Beck, E. Liebscher, L. Toan, and S. Raghavan, ECS Transactions, 41, 5, pp. 93-99 (2011)**
- **S. Kumari, M. Keswani, S. Singh, M. Beck, E. Liebscher, L. Toan, and S. Raghavan, Effect of Dissolved CO<sub>2</sub> in De-ionized Water in Reducing Damage During Megasonic Cleaning in MegPie, 220<sup>th</sup> ECS Meeting, Boston, MA, (Oct 9-14, 2011)**

# Cost and Environmental Benefits of Biosorption of Copper from CMP Wastewater versus Ion Exchange

Lisa A. Jones, Dr. Kimberly Ogden, Mayrita Arrandale

## Objective

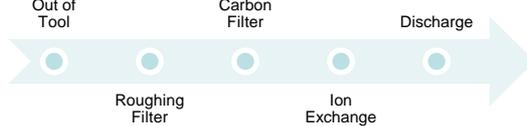
Remove copper (Cu) from chemical mechanical planarization (CMP) wastewater with lipid extracted algae (LEA) from *Chlorella protothecoides*, a freshwater algae

### Background

- Typical semiconductor facility or “fab” manufacturing 300 mm wafers produces approximately 57 liters of wastewater per wafer containing approximately 20 ppm copper
  - Fab will generate over 17 million gallons of wastewater each year with 25,000 wafers per month out put
- Algae is being grown in record quantities for biodiesel; LEA is waste

### CMP Effluent Flow Diagram

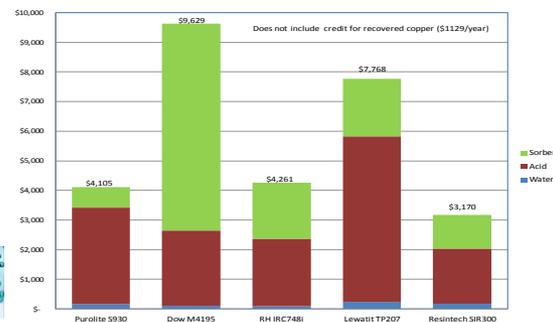
MCLG for copper is 1.3 mg/L or 1.3 ppm set by EPA



### Assumptions

- Study is based on treating 32.5 lpm (8.6 gpm) of wastewater containing 20 ppm copper and a pH of 3.0
- Recovery of the copper bound to these resins can be achieved by backwashing with 10% by weight sulfuric acid
- Estimates assume copper removal requires approximately 6 cubic feet of resin with a design flow of 1.5 gallons per minute per cubic foot: bed depth of 1 meter

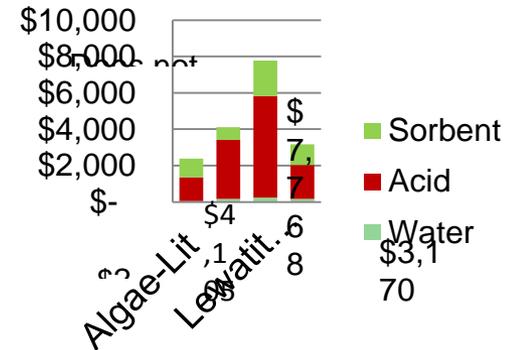
## Annual Ion Exchange Treatment Costs



### Standard Costs

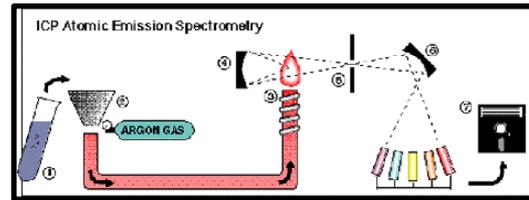
Resin	IRC 748	TP 207	SIR 300
Supplier	Rohm & Haas	Lewatit	Resintech
Bulk density, g/L	725	674	750
copper loading, g/l	35	39	43
Life, yrs	6	6	2
Cost, \$/ft <sup>3</sup>	345	3,494	630
Acid Rinse, BV	3.5	3	6
Water Rinse, BV	9	6	12

## Comparison of Annual Treatment Costs Algae vs. Resin



### ESH Metrics and Impact

- Reduction in emission of copper in wastewater from semiconductor manufacturing plants
- Current emissions are ~ 2 to 20ppm from CMP
  - *Chlorella* sp. capable of removing ~10ppm Cu from water
- Reduction in chemicals and goods currently used to remove copper from CMP waste streams
- Current methods use activated carbon and ion exchange resins to remove copper that require charging with acids: 10% by weight sulfuric acid



### Inductively Coupled Plasma (ICP) Process Cu(II) Detection Method

## ICP Results

Sample	Cu Adsorbed, q (mg/g)
Algae 119-122	9,990
Algae 114-116	8,112
PurLite	9,941
Lewatit	9,945
ResinTech	9,947



### Conclusion

- Algae is an effective option to remove Cu(II) from CMP Waste
- *Chlorella* sp. is able to adsorb same quantity of Cu as resins

### Publications, Presentations, and Recognitions/Awards

Science Foundation Arizona Fellow 2008-2010 & NSF GK-12 Fellow 2010-2011

NSF-CMMI Research and Innovation Conference Grantee

Jones L, Ogden K. (2012). Silica & Titania Nanoparticles Impact on Water Quality: Experiments Involving *Ralstonia pickettii* in Nutrient Rich and Poor Media. Environmental Progress, in publication.

# Tribological, Thermal, and Kinetic Attributes of 300 vs. 450 mm CMP

## Senior Researchers:

- Ara Philipossian, Professor, PI
- Yun Zhuang, Postdoctoral Fellow
- Yasa Sampurno, Postdoctoral Fellow

## Sponsored by:

- Intel
- Cabot Microelectronics
- Fujimi

## Students

- Yubo Jiao, Xiaoyan Liao, and Changhong Wu, Ph. D. candidates, Chemical and Environmental Engineering, UA

## Highlights:

We established 450-mm wafer processing capability and results showed that CMP consumable consumption can be reduced more than 20% through processing larger size (450 mm) wafers due to the associated higher material removal rate.

# **Aggressive Diamond Characterization and Wear Analysis during Chemical Mechanical Planarization**

## **Senior Researchers:**

- **Ara Philipossian, Professor, PI**
- **Yun Zhuang, Postdoctoral Fellow**
- **Yasa Sampurno, Postdoctoral Fellow**

## **Sponsored by:**

- **Cabot Microelectronics**

## **Students**

- **Changhong Wu, Yubo Jiao, and Xiaoyan Liao, Ph. D. candidates, Chemical and Environmental Engineering, UA**

**In this project, we aimed to extend pad conditioner life by identifying its aggressive diamonds and investigating aggressive diamond wear evolution during chemical mechanical planarization.**