



Outlook For Resist Design at 157 nm

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Semiconductor Manufacturing



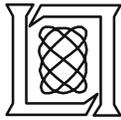
Outline

- **Overview: Why 157 nm?**
- **Challenges for 157-nm resists**
 - **Difficulties in using existing resists**
 - **VUV absorbance of common polymers**
 - **Requirements for new materials**
- **Summary**



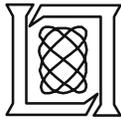
Interest in Developing 157-nm Lithography is Growing

- **Sematech's "Next Generation Lithography" (NGL) Conference on Dec. 10, 1998 revealed that many semiconductor companies are concerned about the timing of the NGL choices (EUV, EPL, IPL, X-Ray)**
- **157-nm lithography emerged as an interim choice necessary to bridge the gap to the NGLs**
- **Recent advances in lasers and optical materials has encouraged the industry that 157-nm is technically feasible (although cost-of-ownership still remains a concern)**



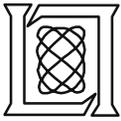
The 157-nm Challenge

- **Potential technical “show stoppers” are materials related:**
 - Pellicle materials (inorganic membranes, fluoropolymers)
 - Mask materials (fluorine-doped synthetic silica)
 - Resist materials (hydrofluoropolymers, siloxanes, etc.)
- **Resist challenges are multifaceted**
 - Transparency (use of thin layers)
 - Photochemistry (both desired and undesired)
 - Solubility and adhesion
 - Outgassing and contamination



The 157-nm Resist Challenges

- **Using existing resists requires ultrathin resists**
 - Defect levels in sub-100-nm resist layers may be an issue
 - Problem is similar to that faced by EUV resist efforts
- **VUV photochemistry of existing resists not well studied**
 - High G_x values can lead to reduced contrast
 - High G_s values can cause outgassing
- **New routes: Little data exists for functional-group contributions**
 - Absorbance must be weighed against other properties
Solubility/etch resistance/transparency a three-way balance
 - Preliminary absorbance studies suggest possible routes

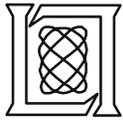


Ideal Scenario Would Allow Use of Existing Resists

- **Absorbance at 157 nm requires using thin resists**
 - **Thin layers have more defects**
 - Little data exists - risk unknown
 - How thin is “thin”?
 - **Etch resistance questionable (hard masks offer solution)**
 - **EUV ultra-thin resist studies can provide useful data**
- **New photochemical pathways imperil design chemistry**
 - **Quantum yields of unwanted side reactions may begin to compete with photoacid generation (7.9 eV photon)**
 - **Undesired photochemistry can:**
 - Increase the molecular weight which in turn lowers the dissolution rate and reduces the contrast
 - Decrease the molecular weight which can lead to volatile fragments and outgassing
 - Generate free radicals which may activate new chemical pathways that affect acid generation efficiency



Difficulties in Using Existing Resists

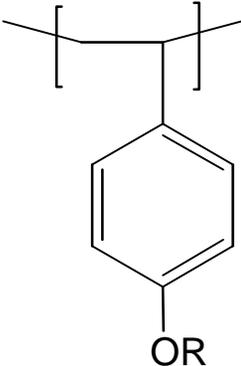
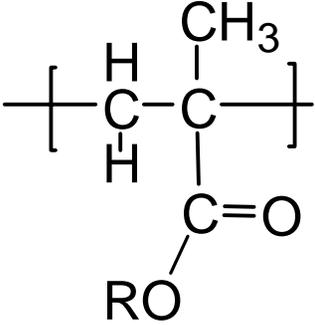


Aspects of Using Existing Resists in Ultrathin Layers

- Are any current platforms more transparent at 157 nm?
- No manufacturing experience to assess defect-related yield loss or film defect metrology. Do pinholes even transfer?
- The whole resist is an interface!
 - substrate and atmospheric interactions critical
 - surface inhibition effects can have larger impact on performance than with thick resist
- Resist thinning must be better controlled
 - 2% thinning in a 700-nm resist becomes 20% in a 70-nm resist!
 - 20% thinning combined with limited aerial image will yield badly rounded profiles
- Etch resistance may be limited even for hard-mask pattern transfer

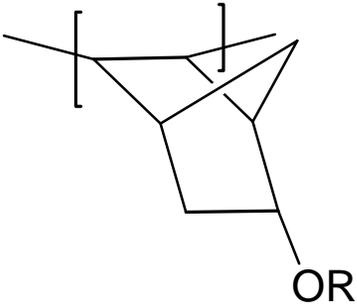
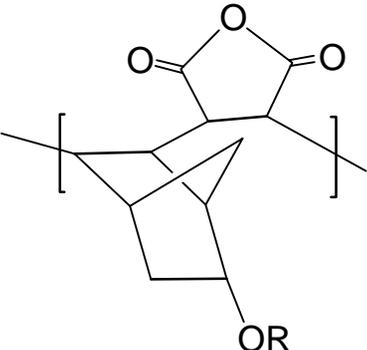


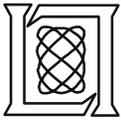
All Existing Polymer Platforms Require Thicknesses <70 nm

	POLYMER	A / μm
Phenolic		Poly(styrene) 6.2
		Poly(hydroxy styrene) 6.3
		Commercial DUV resist A 7.3
		Commercial DUV resist B 7.7
		Commercial DUV resist C 8.3
Acrylic		Poly(methyl methacrylate) 5.7
		Poly(norbornyl methacrylate) 6.7
		Poly(adamantyl methacrylate) 6.7
		V1.1 terpolymer resin 7.4
		Poly(isobornyl methacrylate) 7.6
		V1.1 terpolymer resist 8.7
		Poly(acrylic acid) 11.0

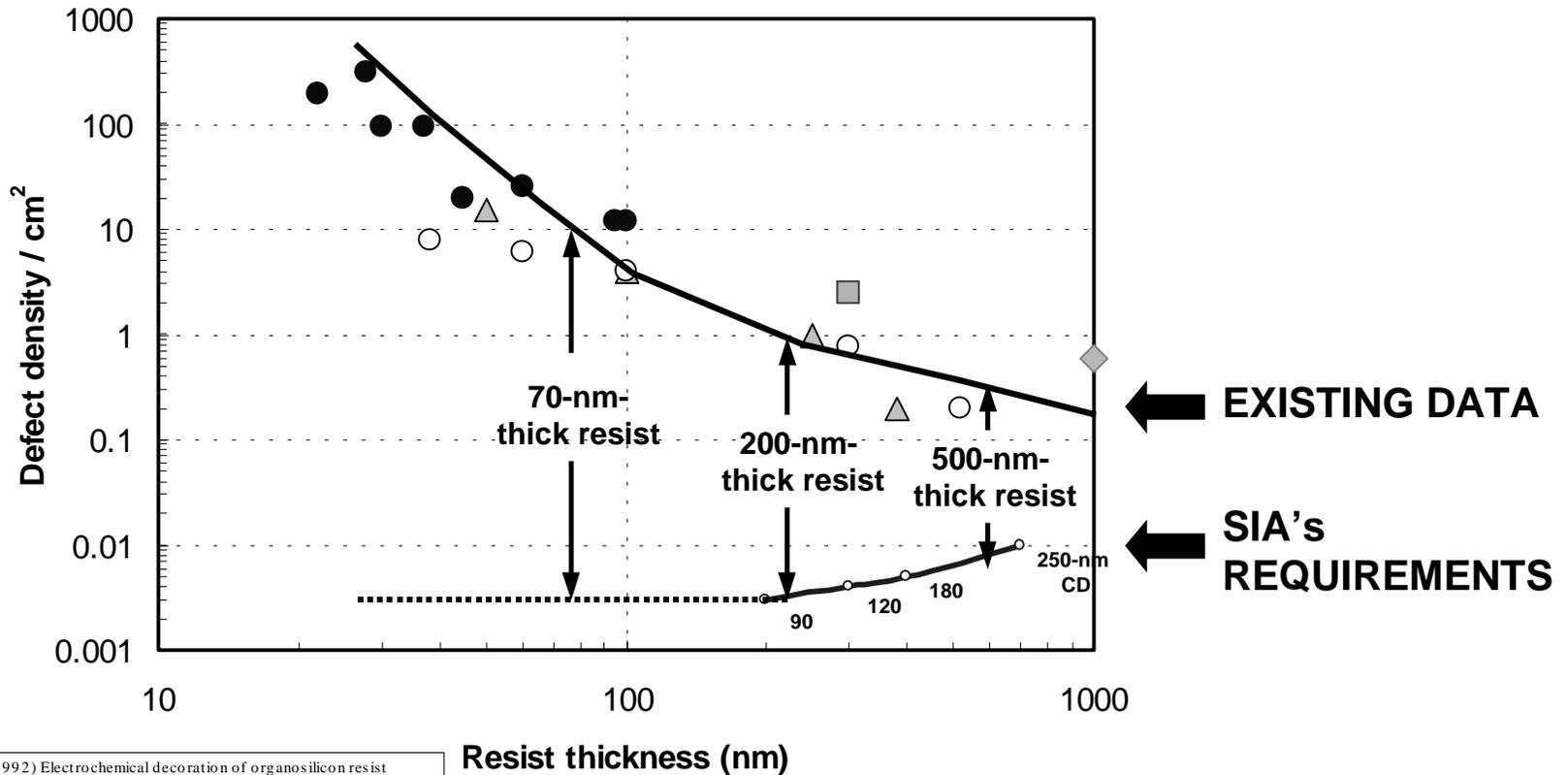


All Existing Polymer Platforms Require Thicknesses <70 nm

	POLYMER	A / μm	
Norbornene		Poly(norbornene) (source A)	6.2
		Poly(norbornene) (source B)	6.8
		ROMP Poly(norbornene)	6.8
		Poly(norbornene) (source C)	7.1
Norbornene Maleic Anhydride		Commercial resist A	7.3
		Poly(norbornene-alt-maleic anhydride) A	8.2
		Poly(norbornene-alt-maleic anhydride) B	9.0



The Risk Of Using Ultrathin Resists: Defects!

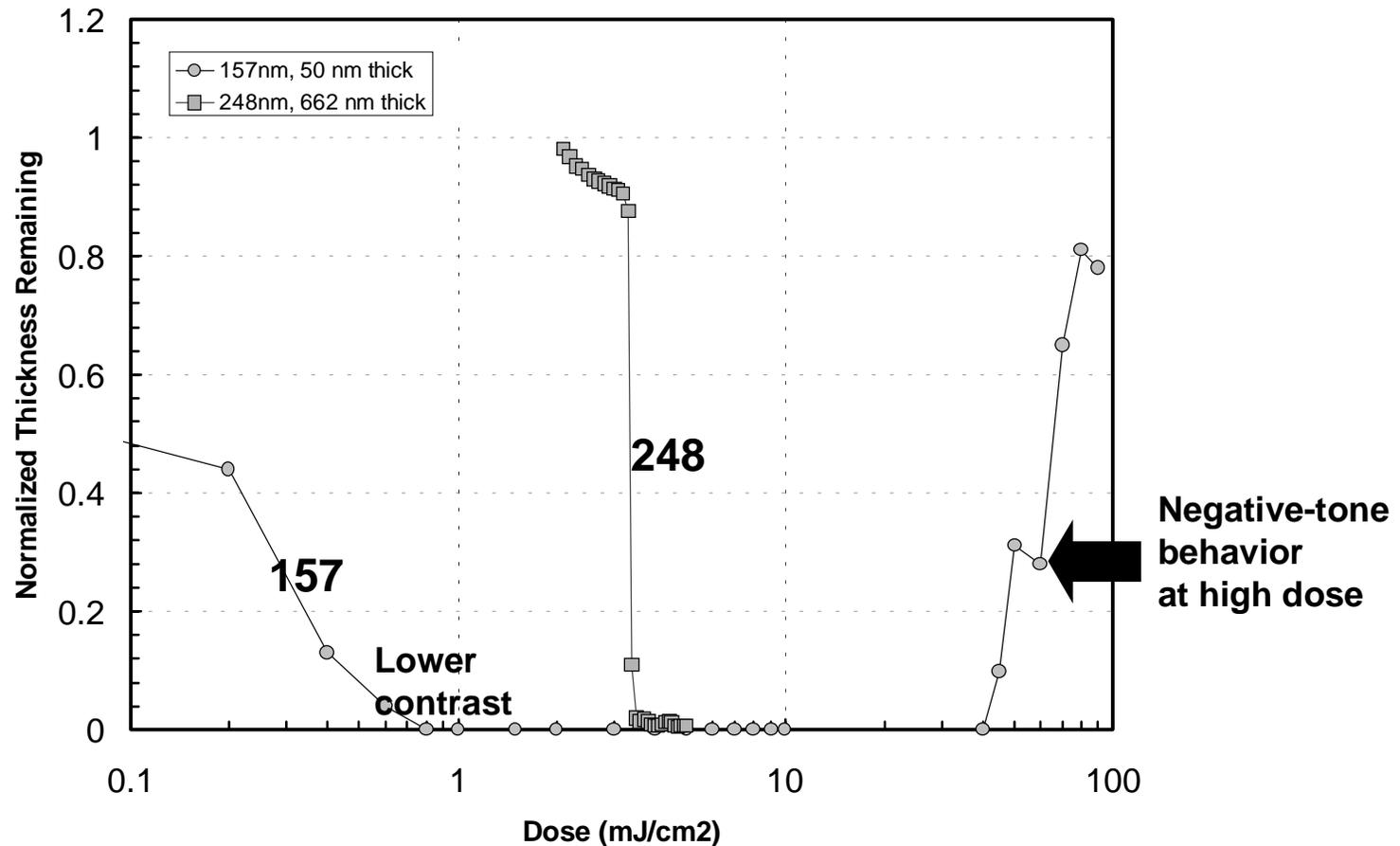


- Kunz, MITLL (1992) Electrochemical decoration of organosilicon resist
- Muller, IBM (1992) Plasma decoration of organosilicon resist
- ▲ Early, AT&T (1992) Wet-etch decoration of organic resist
- ◆ Muller, IBM (1992) Plasma decoration of organic resist
- Kuan, Stanford (1988) Wet-etch decoration of organic resist
- 1997 SIA Roadmap

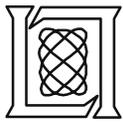
Risks associated with use of ultrathin resists in manufacturing are unknown.



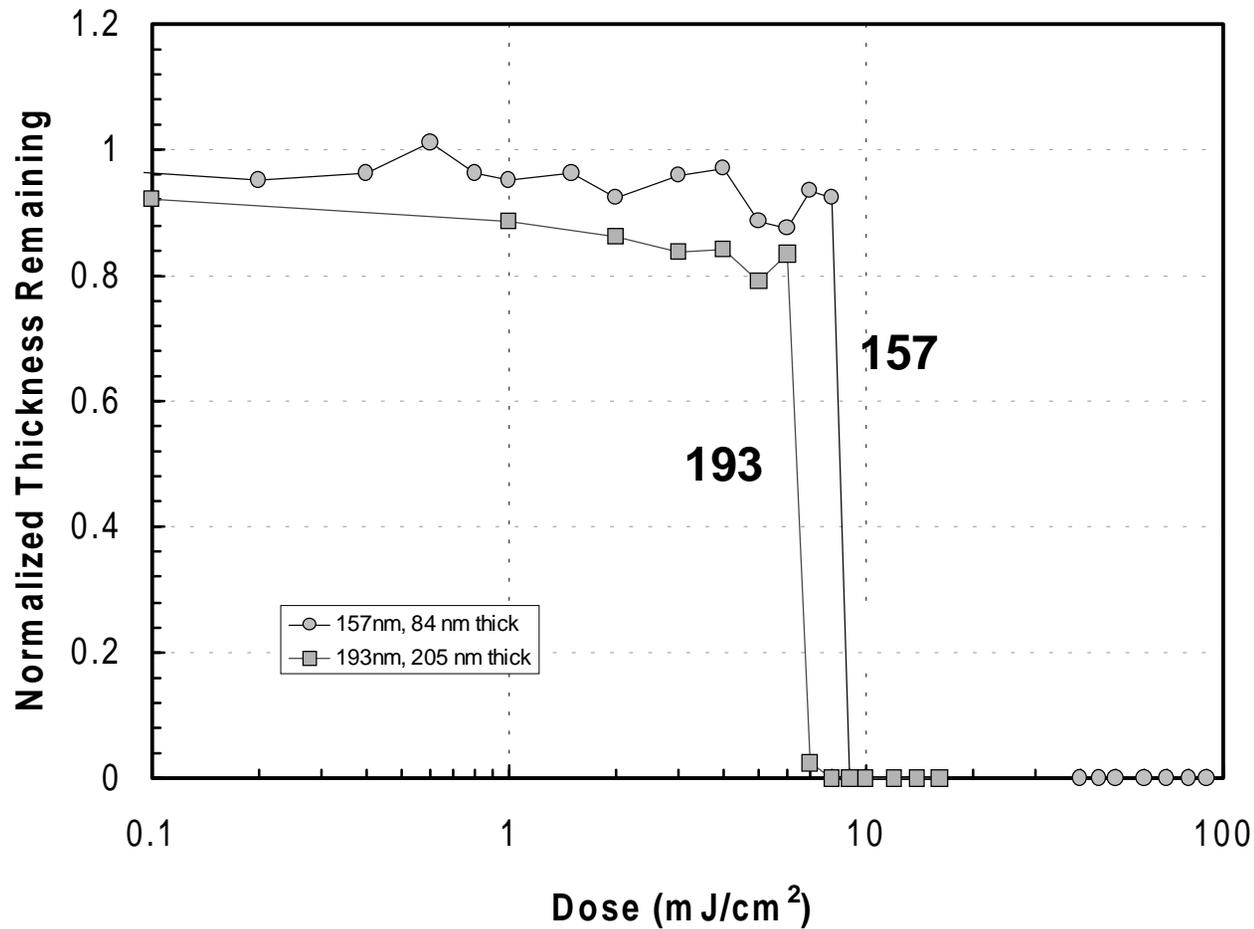
Performance of a DUV Phenolic Resist: 157 nm vs. 248 nm

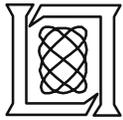


Even if thin layers are tolerable, acceptable lithographic performance is not ensured



Performance of a 193-nm Acrylic Resist: 157 nm vs. 193 nm



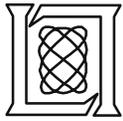


Summary of contrast curve study

Seven resists tested to date:

Resist	Type	Polymer	E_0 (mJ/cm ²)	E_0^{157} (mJ/cm ²)	γ	γ_{157}	Thinning (nm/min)	Neg dose (mJ/cm ²)
A	248	PHOST	3.4	0.8	79	1	48	50
B	193	Acrylic	14	9.0	7	10	30	>90
C	193	CO-MH-Ac	8.0	2.5	9	3	54	>90
D	193	Acrylic	7.5	9.0	8	12	18	>90
E	193	CO-(MH)	4.0	9.0	6	8	4	>90
F	193	Acrylic	7.0	2.0	2	3.3	18	>90
G	ebeam	PHOST	NA	4.0	NA	2	NA	NA

- All resists tested (at 248, 193, or 157 nm) on bare silicon with their thicknesses adjusted to yield film ODs of ~0.4.
- All development times shortened to 15 seconds.



Use of Existing Resists

How it Might Look Based on Current Data

- **Cyclo-olefin or acrylic 193-nm resist**
- **Absorption coefficient requiring a thickness of 55 nm**
- **Sensitivity $<10 \text{ mJ/cm}^2$**
- **Contrast >8**
- **Yet unresolved:**
 - **Etch resistance**
 - **Outgassing (thin layer may help some)**
 - **Reflection control**
 - **Defects**
- **Unresolved issues not critical to “tool testing resist”**
- **Inorganic ARC/hardmask would address etch resistance and reflection control**
- **Improved silicon-containing bilayers may also help**



Photoacid Generator Absorbance

Photoacid Generator (PAG)	Added absorbance to a 200-nm-thick resist with 4 wt% PAG
Naphthylimidotriflate	0.05
Triphenylsulfonium perfluorobutane sulfonate	0.17
Bis(t-butylphenyl)iodonium camphor sulfonate	0.14

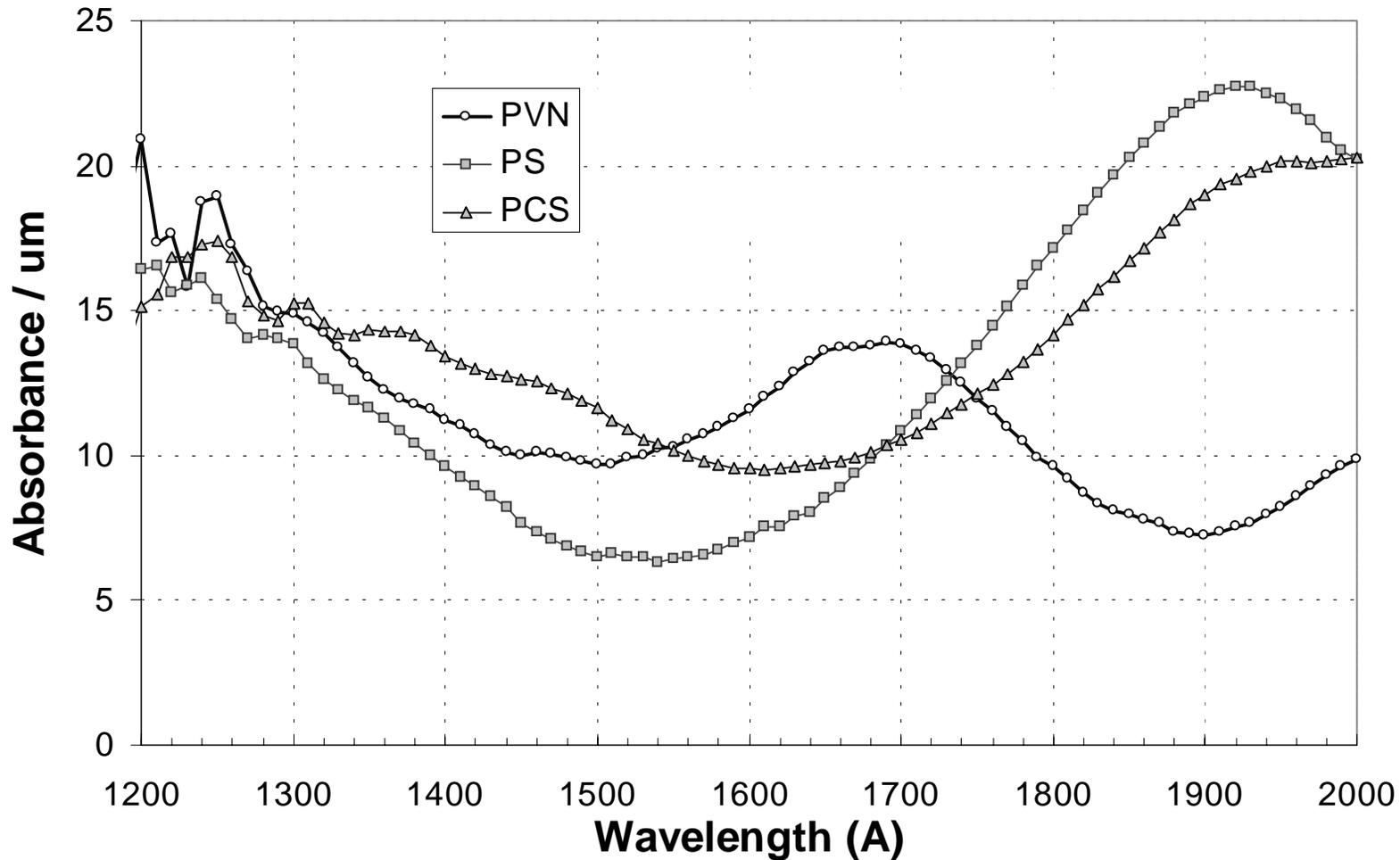
Note: Aromatic PAGs are more transparent at 157 than at 193.



VUV absorbance of common polymers

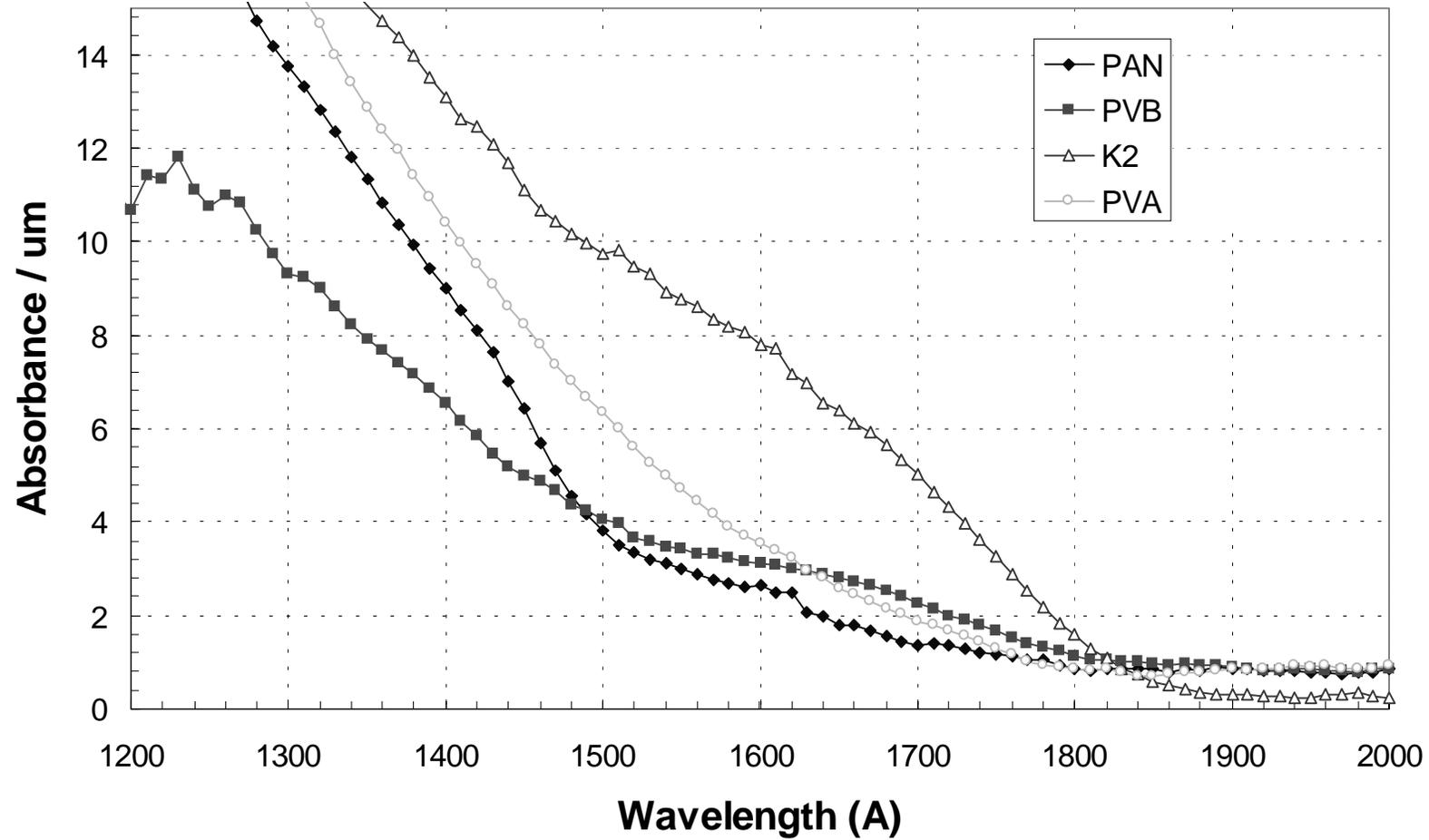


Poly(vinylaromatics): Absorption-band shifting for 157 nm?



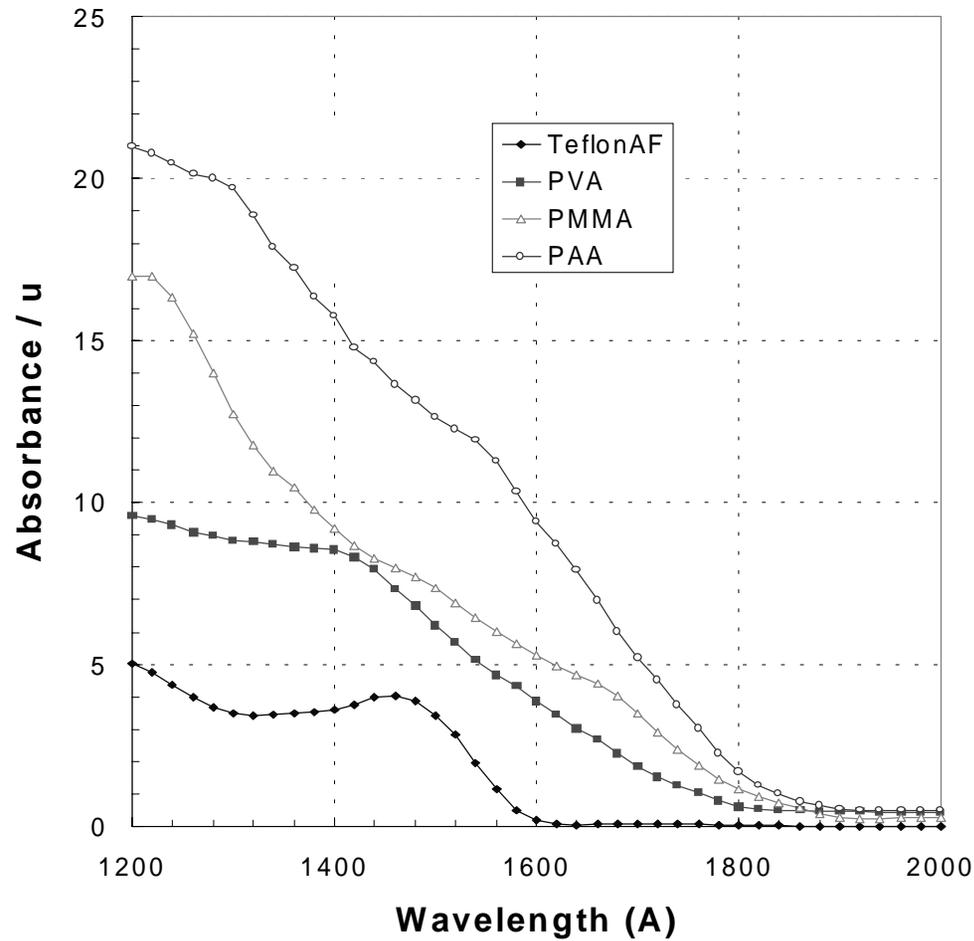


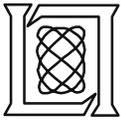
Aliphatic polymers



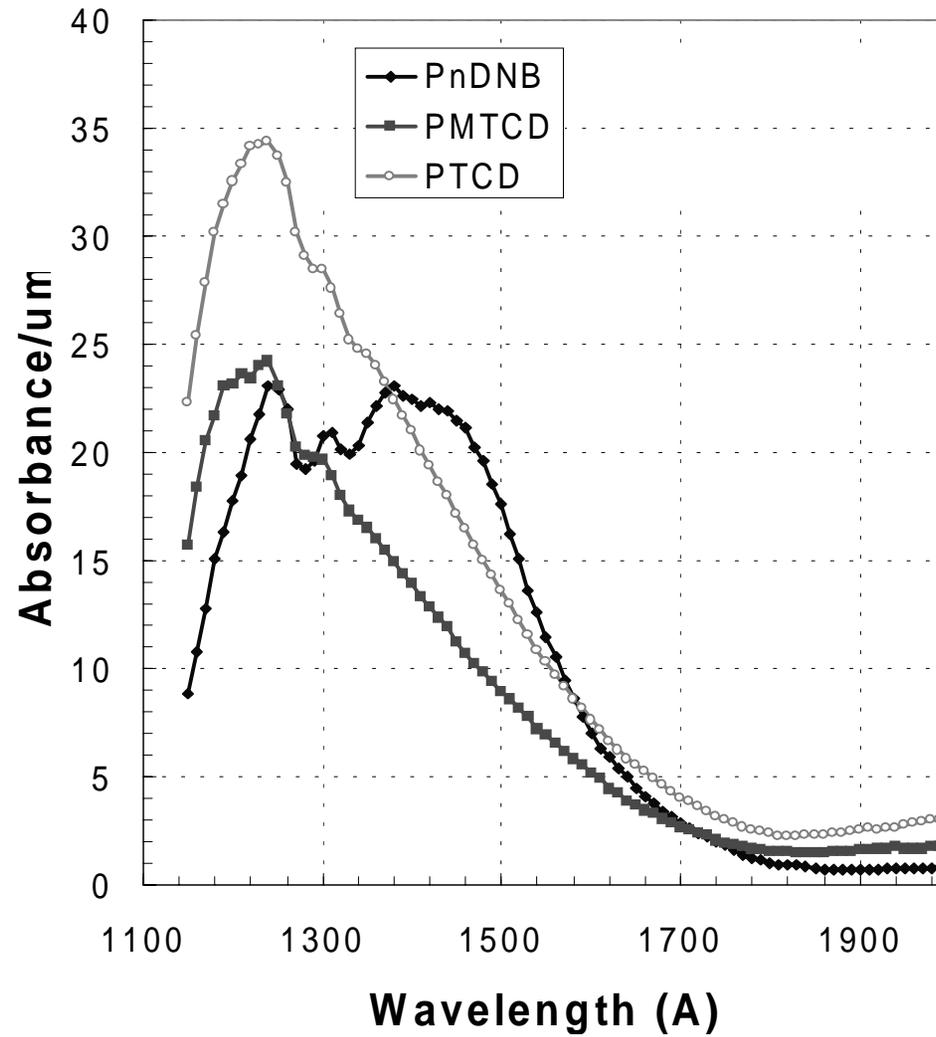


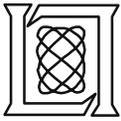
Aliphatic polymers



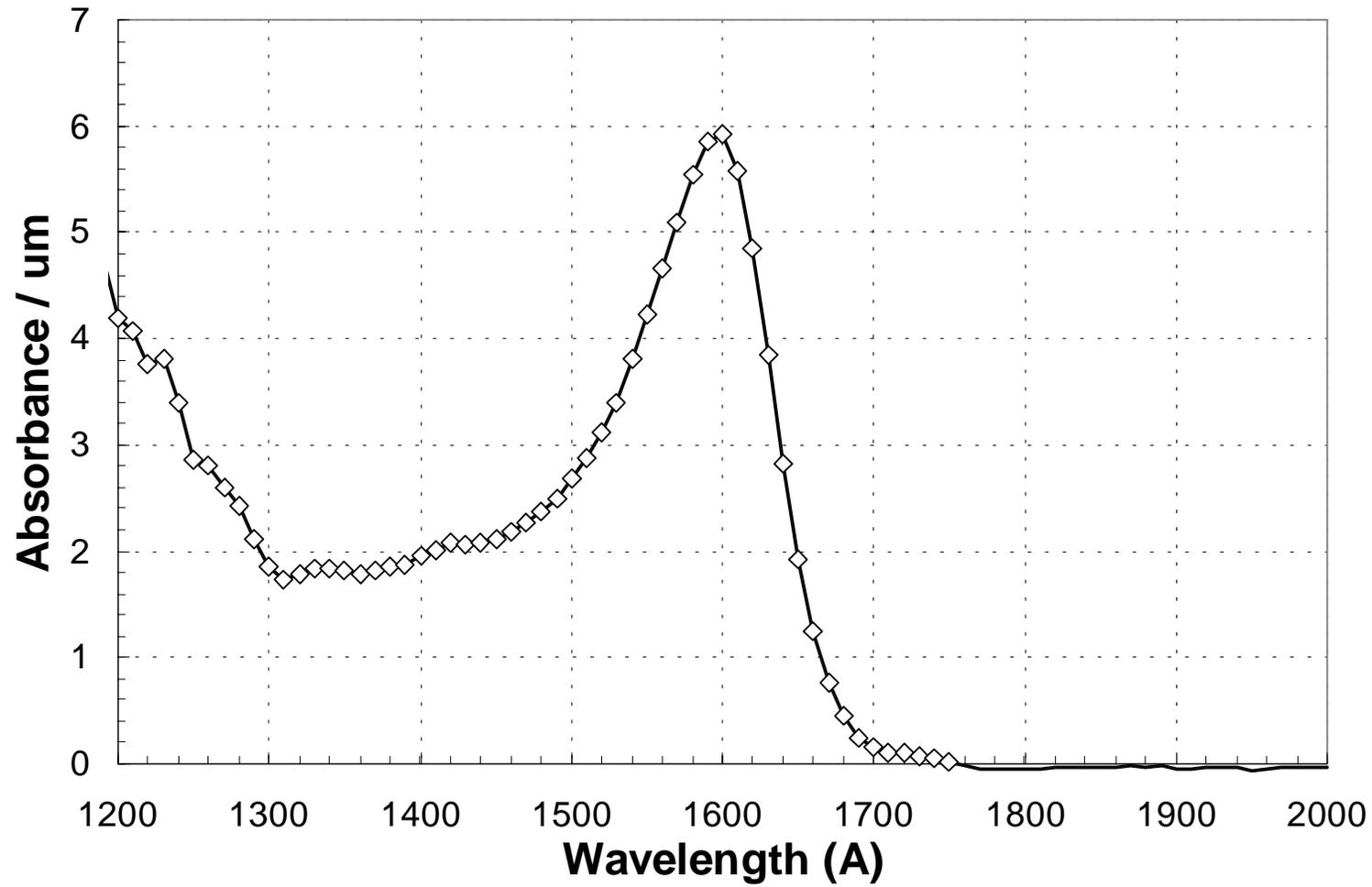


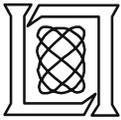
Cyclo-olefins



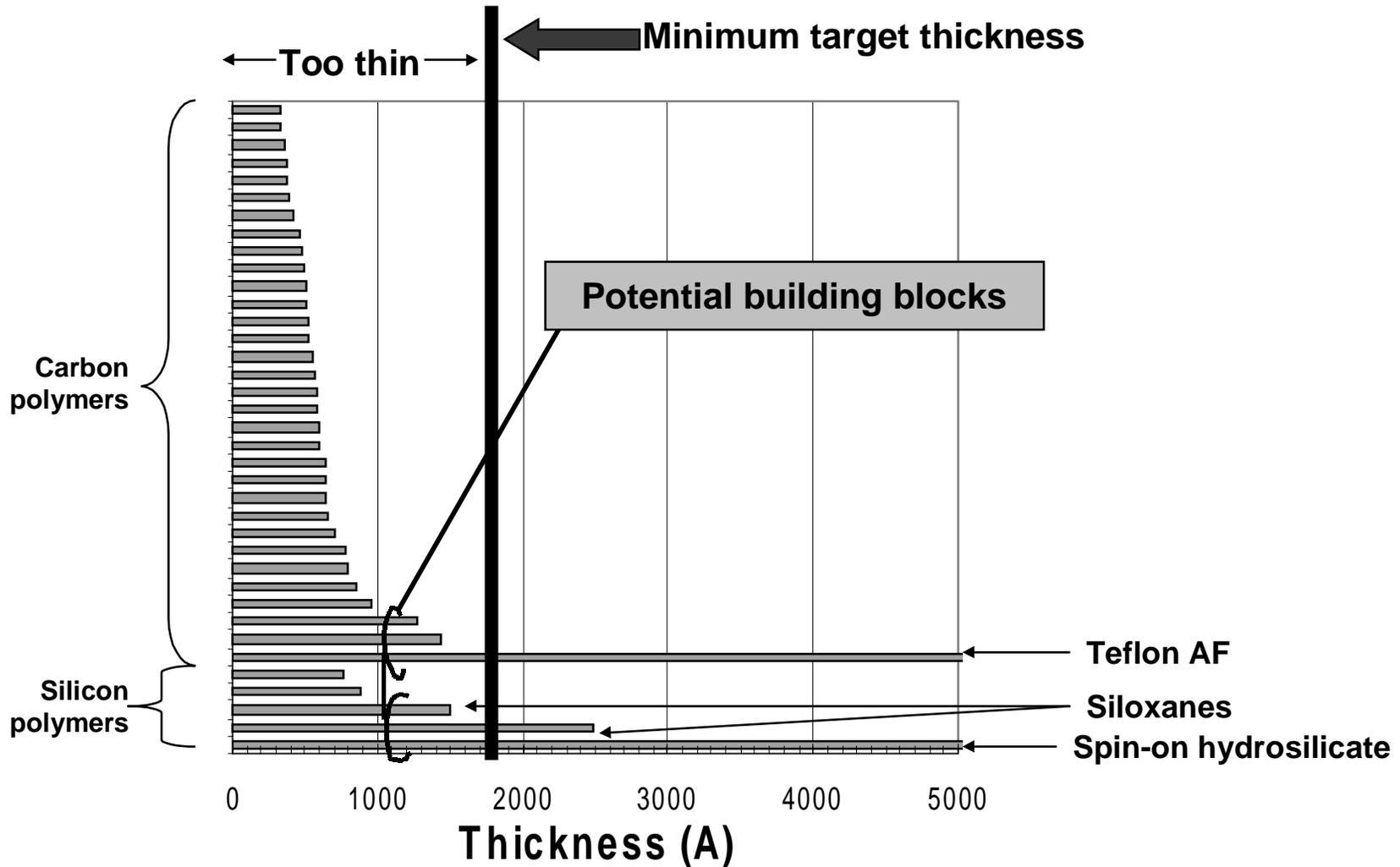


Nafion



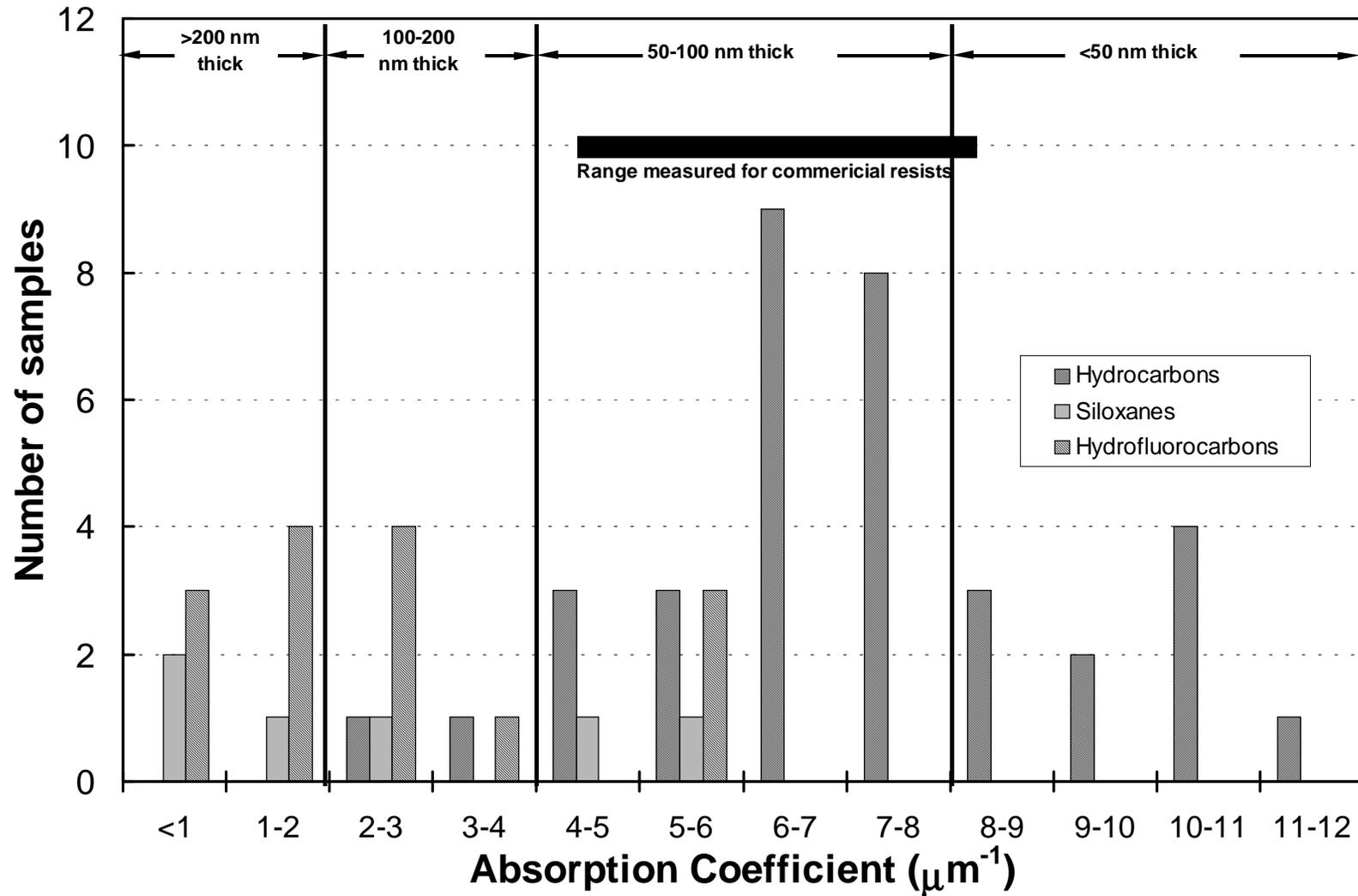


Polymer thicknesses required for an optical density of 0.4





Absorption Summary For Polymers Tested



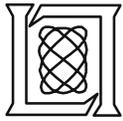


What prospects for new materials have been identified?



Routes to More Transparent Polymers

- **Carbon polymers can be made more transparent by “dilution” via transparent comonomers or side chains**
 - siloxane functionality is best candidate, although fluorocarbons and polyethers are possible as well
- **Avoid all π electrons (i.e., no double bonds)**
- **C-C σ -bond can be made more transparent through addition of transparent electron withdrawing groups (i.e. fluorine)**
 - large impact on dissolution and possibly adhesion
 - transparencies yielding films >200 nm are possible
- **C-C σ -bond can also be made more transparent (to a limit) by controlling the bond conformation**
 - little impact on dissolution and adhesion
 - limited gains in absorbance
- **Judicious manipulation of these approaches should yield a material that can be used in thicknesses of ~200 nm**



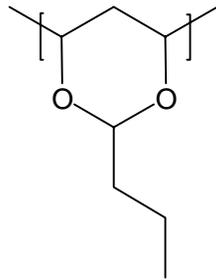
Modification of Existing Platforms With Transparent Functionality

RESIN	A / μm	THICKNESS (OD=0.4)
Methacrylate Resist A	7.5	53
Methacrylate Resist B	8.7	46
Siloxane-modified methacrylate resist	4.6	87
Perfluoroalkyl methacrylate polymer	4.7	85

- **In order to compensate for the carbonyl absorbance, the substituents must be very transparent, i.e. CF_x and SiOR. This alters solubility significantly.**
- **Only modest gains in thickness possible (this applies to all existing resist platforms).**



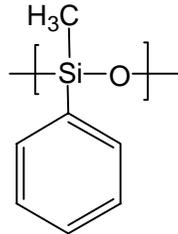
Examples of Transparent Materials



Poly(vinyl butyral)

A = 3.2/μm

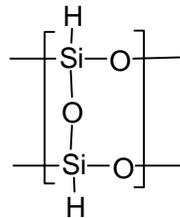
L = 125 nm (OD=0.4)



Poly(phenylmethyl siloxane)

A = 2.7/μm

L = 148 nm (OD=0.4)



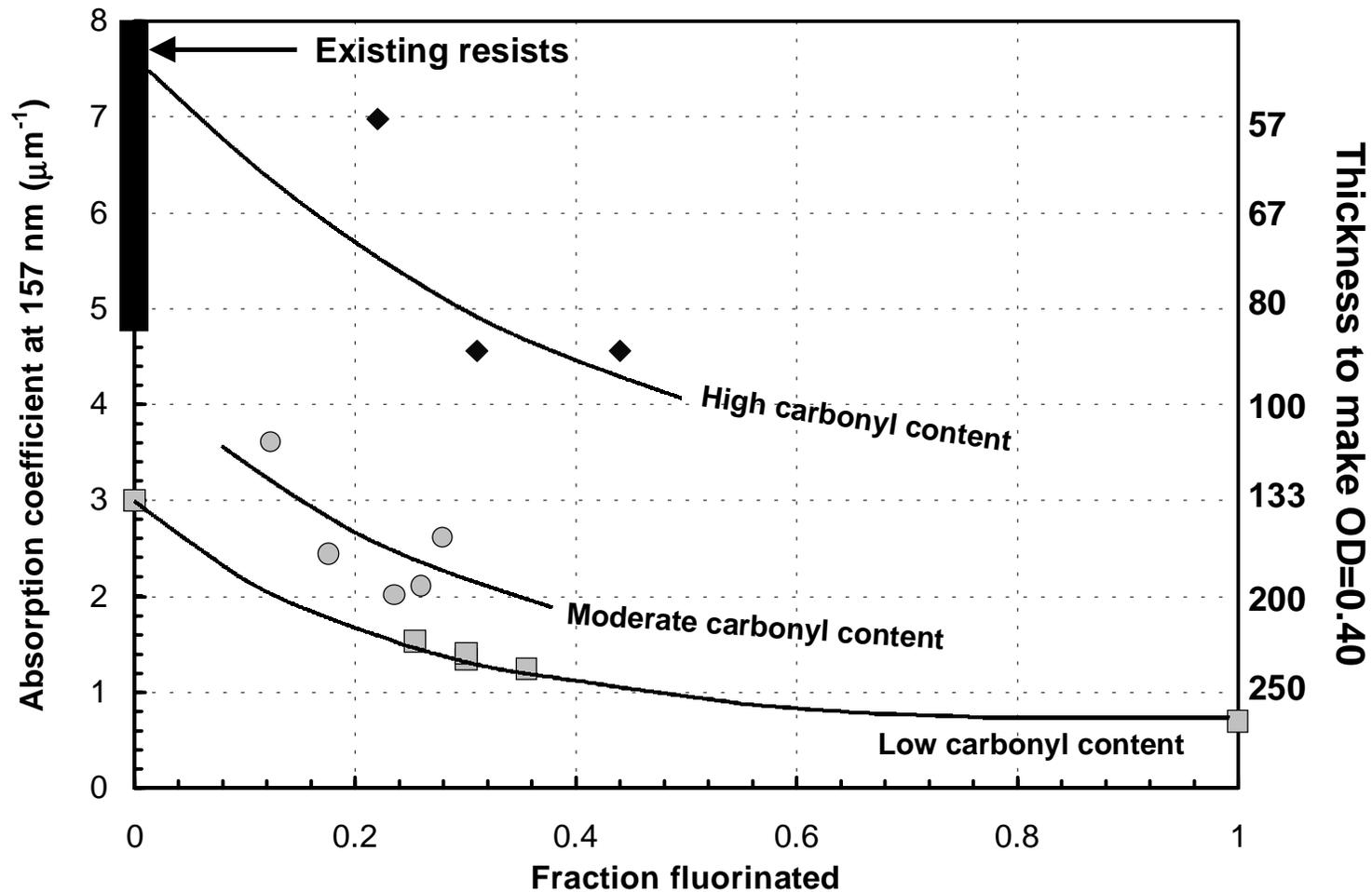
Poly(hydrosilsesquioxane)

A = 0.02/μm

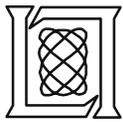
L = 20 μm (OD=0.4)



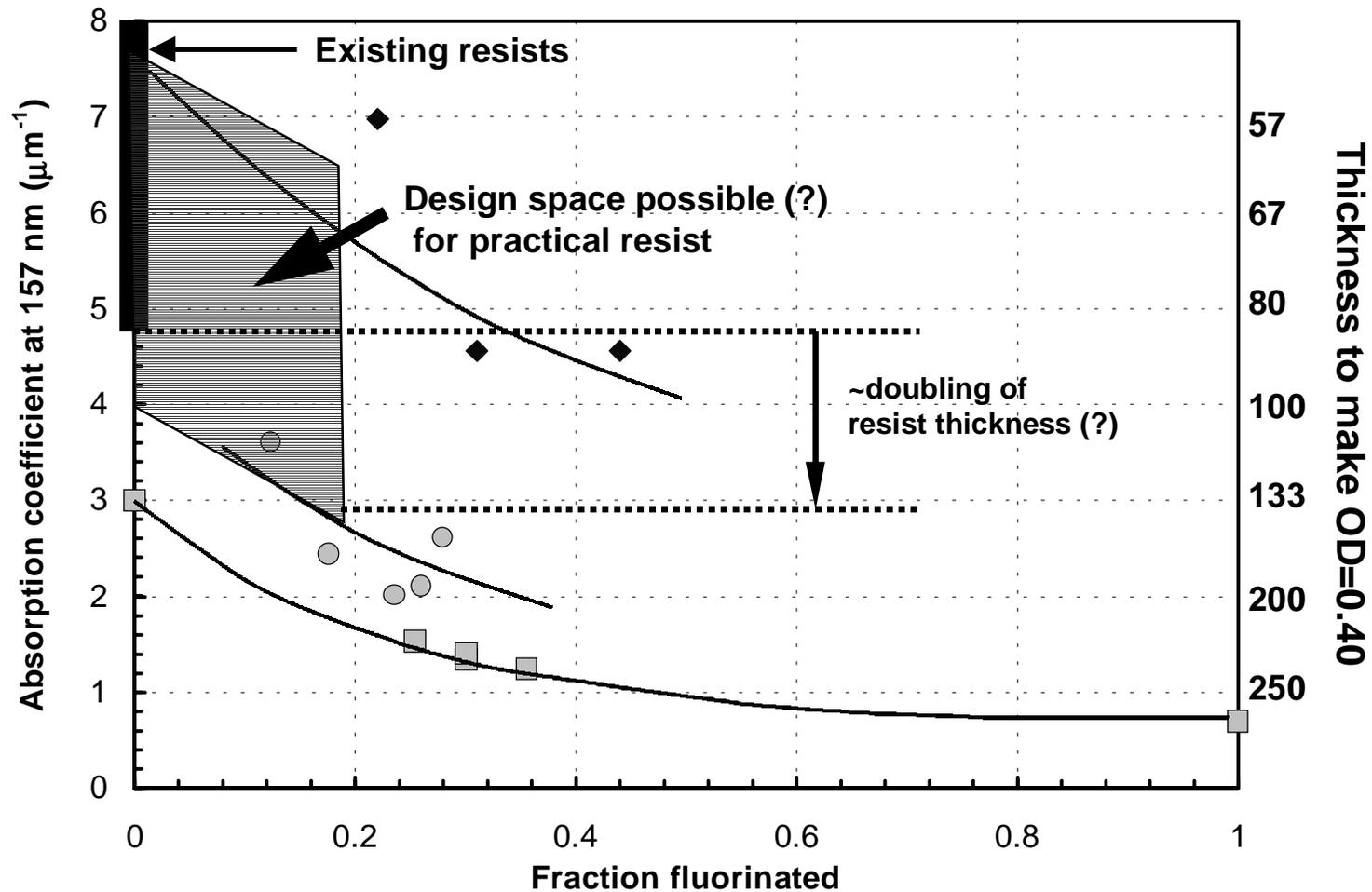
Strong Levers On Absorbance: Fluorocarbon And Carbonyl Content



(Lines drawn to guide the eye)



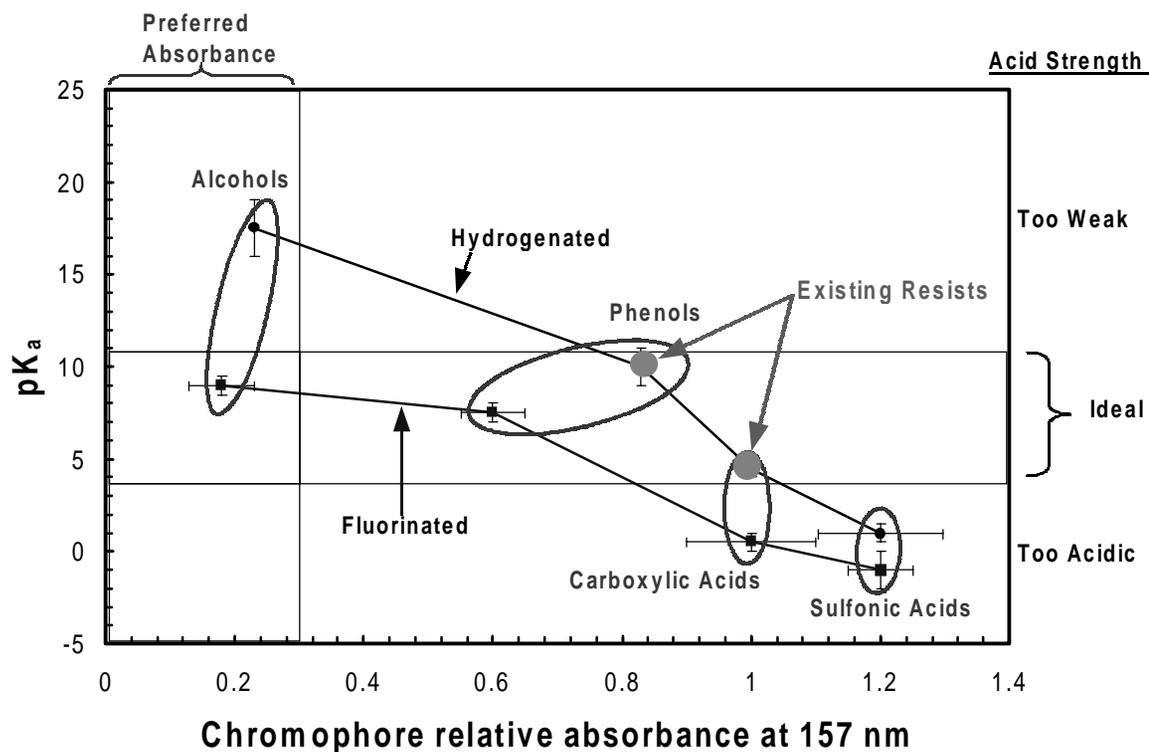
Design Space Possible (?) By Optimal Manipulation Of Fluorine And Ester Content



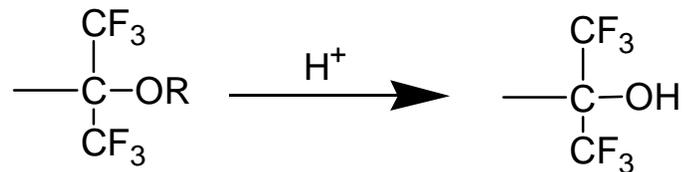
(Lines drawn to guide the eye)



Fluorine-Activated Alcohols Offer Route to Transparency and Base Solubility



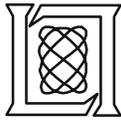
π -electron-free base solubilizing group
(Hoechst, 1992)





Routes to 157-nm Resists

- Existing hydrocarbon resists will be applicable for thicknesses < 80 nm
- Slight modifications to existing resists should afford thicknesses between 80 and 110 nm
 - Siloxyl-functionalized resist an obvious route to bilayer resist
 - Simply putting hexafluoropropanol-protected groups on existing resists should yield transparencies in this area
- No base soluble and transparent homopolymers exist to obtain resist thicknesses > 100 nm
 - Copolymerization is required to obtain desired resin
 - A combination of approaches will be necessary for >150 nm
Siloxyls, hexafluoropropanols, copolymerization, etc.



Outgassing as a Design Concern

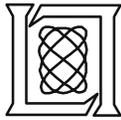
- **Quantum yields for volatile product formation have been measured for many polymers at 253.7 nm**
 - poly(t-butyl acrylate) has $\Phi=0.083$ for isobutene liberation*
 - isobutene primary product observed in t-butyl ester resists (assisted by acidolysis)
 - rates at 157 nm may be even higher
- **Open frame exposure experiments conducted at Lincoln have shown some t-butyl ester resists measurably thin during exposure to 157 nm (unlike 193 and 248 nm)**
- **The Lincoln/Sematech resist outgassing CRDA has shown that deposition of outgassed products occurs more efficiently in a nitrogen ambient, which is unique to 157 nm**
- **Resist-related contamination should constitute an up-front concern to the tool and resist designers**

*Reinisch and Gloria, 1968



Summary

- **An existing commercial resist should be suitable as a tool-testing resist**
 - Selection criteria based on combination of absorbance and lithographic performance at 157 nm
 - Maximum thickness ~90 nm
 - Further work (defects, etch resistance, integration) needs to be done to use in manufacturing, but EUV effort should help
- **Opportunities exist to develop new materials**
 - Absorbance at 157 nm still dominated by molecular transitions, and hence new chemistries can afford greater transparency
 - It appears to be a balancing act between absorbance, solubility characteristics, and etch resistance
 - C-F and Si-O functionality should play key roles
 - Large synthesis effort needed to make headway (including even basic structure-property studies)



Acknowledgments

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